# SYLLABUS

**OF** 

B. TECH

**OF** 

I YEAR

B. TECH. (CBCS)

DEPARTMENT OF CIVIL ENGINEERING

INTEGRAL UNIVERSITY LUCKNOW

## SYLLABUS AND EVALUATION SCHEME

## **Branch: B. Tech Civil Engineering Program**

(w.e.f. 2020-21)

## Year - I, Semester - I

					Per	iods		E					
S. No.			Name of Subject	L	Т	P	C	_	ontinu ssessn (CA	nent )	Exam ESE	Subject Total	
								UE	TA	Total			
1	BS	CH101	Chemistry	3	1	-	4	40	20	60	40	100	
2	BS	MT101	Engineering Mathematics I	3	1	-	4	40	20	60	40	100	
3	ESA	ME101	Basic Mechanical Engineering	3	1	-	4	40	20	60	40	100	
4	ESA	CS101	Computer Programming	3	1	-	4	40	20	60	40	100	
5	BS	ES101	Environmental Studies	2	1	-	4	40	20	60	40	100	
			PRACTICAL / DRA	WI	NG	/ <b>D</b>	ESI	GN					
6	BS	CH102	Engineering Chemistry Lab	ı	-	2	1	40	20	60	40	100	
7	ESA	ME102	Mechanical Engineering Lab	ı	-	2	1	40	20	60	40	100	
8	ESA	CS102	Computer Programming Lab	-	-	2	1	40	20	60	40	100	
9	HM	LN151	Professional Communication Lab	-	-	2	2	40	20	60	40	100	
	Total					8	24					900	

L – Lecture; T – Tutorial; P – Practical; C – Credits; UE – Unit Exams; TA – Teacher Assessment Continuous Assessment (CA) = Unit Exams + Teacher Assessment

**Subject Total** = Continuous Assessment (CA) + End Semester Examination (ESE)

**BS** – Basic Sciences **DC** – Departmental Core **HM** – Humanities **OE** – Open Elective

**DE** – Departmental Elective **ESA** – Engineering Science & Art (Foundation Course & Engineering Courses)

#### SYLLABUS AND EVALUATION SCHEME

## **Branch: B. Tech Civil Engineering Program**

(w.e.f. 2020-21)

## Year - I, Semester - II

1 car	,	ilester .			Per	iods		E	valua	tion Sch	eme	
S. No.	Course Category	Code No	Name of Subject		Т	P	С	Continuous Assessment (CA)			Exam ESE	Subject Total
								UE	TA	Total		
1	BS	PY101	Physics	3	1	-	4	40	20	60	40	100
2	НМ	LN101	Basic Professional Communication	2	1	-	3	40	20	60	40	100
3	BS	MT112	Engineering Mathematics II	3	1	-	4	40	20	60	40	100
4	ESA	EE103	Basic Electrical Engineering	3	1	-	4	40	20	60	40	100
5	ESA	EC101	Basic Electronics	3	1	-	4	40	20	60	40	100
			PRACTICAL / DRA	WI	NG	/ <b>D</b>	ESI	GN				
6	BS	PY104	Physics Lab	-	-	2	1	40	20	60	40	100
7	ESA	EE104	Electrical Engineering Lab	-	-	2	1	40	20	60	40	100
8	ESA	ME103	Engineering Graphics	-	-	2	1	40	20	60	40	100
9	ESA	ME104	Workshop Practice	-	-	2	2	40	20	60	40	100
	Total					8	24					900

 $<sup>\ ^{**}</sup>$  A non credit foundation course, Candidate has to pass the course be securing at least 50% marks up to second semester.

L – Lecture; T – Tutorial; P – Practical; C – Credits; UE – Unit Exams; TA – Teacher Assessment

Continuous Assessment (CA) = Unit Exams + Teacher Assessment

**Subject Total** = Continuous Assessment (CA) + End Semester Examination (ESE)

**BS** – Basic Sciences **DC** – Departmental Core **HM** – Humanities **OE** – Open Elective

**DE** – Departmental Elective **ESA** – Engineering Science & Art (Foundation Course & Engineering Courses)



<b>Effective from Session:</b>	2015-16									
Course Code	CH101	Title of the Course	Chemistry	L	T	P	C			
Year	First	Semester	First	3	1	0	4			
Pre-Requisite	10+2 with Chemistry	10+2 with Chemistry Co-requisite								
Course Objectives	<ul><li>Develop curiosity and interest</li><li>Acquire an appropriate knowl</li><li>Develop an appreciation of ch</li></ul>	edge and understanding in	•							

	Course Outcomes
CO1	Analyze and compare magnetic behavior and stability of hetero-nuclear diatomic molecules, Significance of hydrogen bonding, band theory, radius ratio, density of unit cell, fullerenes and graphite
CO2	Comprehension of types of polymers to make an appropriate choice of use of polymers (Natural, synthetic and biodegradable).
CO3	Compare reaction intermediates and mechanism of chemical reactions and isomerism.
CO4	Interpret phase rule, phase diagram, corrosion and its prevention, calculation of activation energy, rate constant, half-life period, emf of electrochemical cells, construction and operation of galvanic cell and concentration cells,
CO5	Determination of calorific value, analyzing water softening methods, principles, instrumentations of UV, IR and NMR spectroscopy and their applications.

Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
Chemical bonding and state of matter	Molecular theory of hetero diatomic molecules, Band theory of bonding in metals, Hydrogen bonding. Solid state chemistry: Radius ratio rule, Space lattice (only cubes), Types of Unit cells, Bragg's law, calculation of density of unit cell. One and Two Dimensional solids, Graphite as two dimensional solid and its conducting and lubricating properties. Fullerene and its applications	08	CO1
Polymers	Polymerization and its classification, Thermoplastic and thermosetting resins. Elastomers (Buna-S, Buna-N, thiokols, polyurethanes, silicons), Polyamides (Nylon-6, Nylon-6,6, Nylon-6,10, Nylon-11, Kevlar), Polyesters (Terelene), Polyacrylates (PMMA, PAN, PVC). Organic conducting and biodegradable polymers.	08	CO2
Structural and mechanistic concepts in organics	Stability of reaction intermediates, e.g. Carbanions, Carbocations and free radicals. Types of organic reactions, mechanism of nucleophilic substitution reactions. Mechanism of the following name reactions.  1) Aldol condensation 2) Cannizzaro reaction 3) Beckmann rearrangement 4) Hofmann rearrangement and 5) Diels-Alder reaction E-Z Nomenclature. R.S configuration, Optical isomerism of organic compounds containing one chiral center. Examples of optically active compounds without chirality. Conformations of n-butane.	08	CO3
Reaction kinetics, Phase rule, Electrochemistry and Corrosion	Order and molecularity of reactions. First and second order reactions. Energy of activation. Phase Rule, its application to one component system (water). Equilibrium potential, electrochemical cells (galvanic and concentration cells) Electrochemical theory of corrosion and protection of corrosion.	08	CO4
Analytical methods, Fuel and Water treatment	Basic principles of spectroscopic methods. The use of UV, Visible, IR, 1HNMR, for the determination of structure of simple organic compounds. Classification of fuels, determination of gross and net calorific values using Bomb Calorimeter. Hardness of water, softening of water by Lime-Soda process, Zeolites and ion exchange resins process and Reverse Osmosis. Treatment of boiler feed water by Calgon process	08	CO5
	Chemical bonding and state of matter  Polymers  Structural and mechanistic concepts in organics  Reaction kinetics, Phase rule, Electrochemistry and Corrosion  Analytical methods, Fuel and Water	Chemical bonding and state of matter  Molecular theory of hetero diatomic molecules, Band theory of bonding in metals, Hydrogen bonding. Solid state chemistry: Radius ratio rule, Space lattice (only cubes), Types of Unit cells, Bragg's law, calculation of density of unit cell. One and Two Dimensional solids, Graphite as two dimensional solid and its conducting and lubricating properties. Fullerene and its applications  Polymers  Polymers  Polymers  Polymers  Polymers  Polymers  Polymers  Polymerization and its classification, Thermoplastic and thermosetting resins. Elastomers (Buna-S, Buna-N, thiokols, polyurethanes, silicons), Polyamides (Nylon-6, Nylon-6,6, Nylon-6,10, Nylon-11, Kevlar), Polyesters (Terelene), Polyacrylates (PMMA, PAN, PVC). Organic conducting and biodegradable polymers.  Stability of reaction intermediates, e.g. Carbanions, Carbocations and free radicals. Types of organic reactions, mechanism of nucleophilic substitution reactions. Mechanism of the following name reactions.  1) Aldol condensation  2) Cannizzaro reaction  3) Beckmann rearrangement  4) Hofmann rearrangement  4) Hofmann rearrangement and  5) Diels-Alder reaction  E-Z Nomenclature. R.S configuration, Optical isomerism of organic compounds without chirality. Conformations of n-butane.  Order and molecularity of reactions. First and second order reactions. Energy of activation. Phase Rule, its application to one component system (water). Equilibrium potential, electrochemical cells (galvanic and concentration cells) Electrochemical theory of corrosion and protection of corrosion.  Basic principles of spectroscopic methods. The use of UV, Visible, IR, 1HNMR, for the determination of structure of simple organic compounds. Classification of fuels, determination of gross and net calorific values using Bomb Calorimeter. Hardness of water, softening of water by Lime-Soda process, Zeolites and ion exchange resins process and Reverse Osmosis. Treatment of boiler feed water by Calgon process	Chemical bonding and state of matter  Chemical bonding and state of matter  Molecular theory of hetero diatomic molecules, Band theory of bonding in metals, Hydrogen bonding. Solid state chemistry: Radius ratio rule, Space lattice (only cubes), Types of Unit cells, Bragg's law, calculation of density of unit cell. One and Two Dimensional solids, Graphite as two dimensional solid and its conducting and lubricating properties. Fullerene and its applications  Polymers  Polymers  Polymers  Polymers  Polymers  Polymers  Polymers  Stability of reaction intermediates, e.g. Carbanions, Carbocations and free radicals. Types of organic reactions, mechanism of nucleophilic substitution reactions. Mechanism of the following name reactions.  Structural and mechanistic concepts in organics  Structural and mechanistic concepts in organic substitution reactions. Mechanism of the following name reactions.  1) Aldol condensation  2) Cannizzaro reaction  2) Cannizzaro reaction  3) Beckmann rearrangement  4) Hofmann rearrangement  5) Diels-Alder reaction  E-Z Nomenclature. R.S. configuration, Optical isomerism of organic compounds without chirality. Conformations of n-butane.  Order and molecularity of reactions. First and second order reactions. Energy of activation. Phase Rule, its application to one component system (water). Equilibrium potential, electrochemical cells (galvanic and concentration cells) Electrochemical theory of corrosion and protection of corrosion.  Basic principles of spectroscopic methods. The use of UV, Visible, IR, 1HNMR, for the determination of structure of simple organic compounds. Classification of fuels, determination of gross and net calorific values using Bomb Calorimeter. Hardness of water, softening of water by Lime-Soda process, Zeolites and ion exchange

Jain P. C. and Jain M. 1994. Engineering Chemistry. DanpatRai publishing company Pvt. Ltd., Delhi.

Bahl B.S, ArunBahl and Tuli B.D. 2007. Essentials of Physical Chemistry. S. Chand and Co. Ltd., Delhi.

Industrial Chemistry B.K.Sharma, Goel publishing house.

Fundamentals of Chemistry, R.L. Madan, S.Chand Publications

#### e-Learning Source:

 $https://www.bing.com/videos/search?q=MO+diagram\&&view=detail\&mid=205AE2DEEABF42ACF824205AE2DEEABF42ACF824\&\&FORM=VRDGAR\&ru=\%\,2Fvideos\%\,2Fsearch\%\,3Fq\%\,3DMO\%\,2520diagram\%\,26qs\%\,3Dn\%\,26form\%\,3DQBVR\%\,26\%\,3D\%\,2525eManage\%\,2$ 

https://www.bing.com/videos/search?q=phase+diagram+video&&view=detail&mid=D49B5109D6339097E40BD49B5109D6339097E40B&&FORM=VRDGAR&ru=%2Fvideos%2Fsearch%3Fq%3Dphase%2Bdiagram%2Bvideo%26FORM%3DHDRSC3

https://www.bing.com/videos/search?q=organic+reaction+mechanism&qpvt=organic+reaction+mechanism&FORM=VDRE

https://byjus.com/jee/polymers/

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	POZ	103	PU4	105	100	PO/	108	109	POIU	POII	POIZ	P501	P502
CO1	3	2	0	0	0	0	0	0	0	1	0	2	3	0
CO2	3	2	2	0	0	1	0	0	0	1	0	2	3	1
CO3	3	2	0	0	0	0	0	0	0	1	0	2	3	0
CO4	3	2	0	0	0	0	0	0	0	1	0	2	3	0
CO5	3	2	2	1	1	1	2	0	0	1	0	2	3	2

1-Low Correlation; 2- Moderate Correlation; 3- Substantial Correlation

Name & Sign of Program Coordinator	Sign & Seal of HoD

#### **Mapped SDGs**

6- Clean Water and Sanitation	6 CLEAN WATER AND SANITATION
7- Affordable and Clean Energy	7 AFFORDABLE AND CILAMENERGY
8-Decent Work and Economic Growth	8 DECENT WORK AND ECONOMIC GROWTH



<b>Effective from Session:</b>	2017-18						
Course Code	MT101	Title of theCourse	Engineering Mathematics - I	L	T	P	C
Year	I	Semester	I	3	1	0	4
Pre-Requisite	10+2 Mathematics	Co- requisite					
Course Objectives			mathematics which is necessary for gratroduced will serve as basic tools for sp				

	COURSE OUTCOME						
CO1	Able to calculate rank of matrix, characteristic equation & characteristic roots & use the applicability of Caylay Hamilton						
Theorem to find inverse of matrix which is very important in many engineering application.							
CO2	To develop ability to solve higher derivative, expansion of functions in ascending power of variable & partial derivatives.						
CO3	Develops ability to solve Jacobian, error and approximation and Extrema of the function.						
CO4	Learn the evaluation policy of some special function like gamma & Beta function. & their relation which is helpful to						
CO4	evaluate some definiteintegral arising in various branch of Engineering.						
CO5	Able to determine vector differentiation and integration.						

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Unit I	Introduction, Different types of matrices, Algebraic operations, Elementary row and column transformations, Rank of matrix, Linear dependence, Consistency of linear system of equations, Characteristic equation, Cayley-Hamilton theorem, Eigen values and eigen vectors.	08	CO1
2	Unit II	Leibnitz theorem, Partial differentiation, Homogeneous functions, Euler's theorem, Expansion of functions of one and two variables.	08	CO2
3	Unit III	Jacobian, Approximation of errors, Extrema of functions of several variables, Lagrange's method of multipliers (simple applications).	08	CO3
4	Unit IV	Double and triple integrals, Change of order of integration, Gamma and Beta functions, Applications to area and volume, Dirichlet's integral and its applications.	08	CO4
5	Unit V	Scalar and Vector point functions, Gradient of a scalar function, Directional derivative, Divergence and Curl of a vector, Line, Surface and Volume integrals, Green's, Stoke's andGauss divergence theorems (without proof).	08	CO5

#### **Reference Books:**

A Text Book of Matrices, S. Chand & Co. New Delhi

Calculus and Analytical Geometry, Narosa Publishing House, New Delhi

Higher Engineering Mathematics, Khanna, Publishers, Pvt. Ltd

Advanced Engineering Mathematics, Khanna Publication

#### e-Learning Source:

https://nptel.ac.in/courses/122104018/

https://nptel.ac.in/courses/111104092/

https://nptel.ac.in/content/storage2/nptel\_data3/html/mhrd/ict/text/111104092/lec21.pdf

https://nptel.ac.in/courses/111107108/

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	roi ro.	POZ	PO3	104	103	100	107	PU	109	1010	1011	PO12	P501	PSU2
CO1	3	2	1	2	2	1	0	0	0	1	0	2	1	1
CO2	3	2	1	2	2	1	0	0	0	0	0	2	1	1
CO3	3	2	1	1	1	1	0	0	0	0	0	2	1	1
CO4	3	2	1	2	3	1	0	0	0	1	0	2	1	1
CO5	3	1	1	1	2	1	0	0	0	0	0	2	1	1

1- Low Correlation; 2- Moderate Correlation; 3- Substantial Correlation

Name & Sign of Program Coordinator Sign & Seal of HoD



Effective from Session: 2017-18										
Course Code	ME101	Title of the Course	Basic Mechanical Engineering		T	P	C			
Year	I	Semester	I	3	1	0	4			
Pre-Requisite	NONE	Co-requisite	NONE							
Course Objectives	<ul> <li>Zeroth law of</li> <li>To understand systems.</li> <li>Be able to mo equations.</li> <li>Be able to dra statistically de</li> </ul>	thermodynamics.  I and apply first and seco  del the problem using fre  w Shear Force Diagram of  eterminate beams.	thermal sciences and temperature measurent and law of thermodynamics to various procee-body diagrams and reach to solution by (SFD) and Bending Moment Diagrams (Bloom the basis of knowledge of stress, strain a	esses using MD) t	and : ; equi	real ilibriu				

	Course Outcomes							
CO1	explain basic concepts of thermal sciences and temperature measurement on the basis of zeroth law of thermodynamics.							
CO2	Understand and apply first and second law of thermodynamics to various processes and real systems.							
CO3	Model the problem using free-body diagrams and reach to solution by using equilibrium equations.							
CO4	Draw Shear Force Diagram (SFD) and Bending Moment Diagrams (BMD) for statistically determinate beams.							
CO5	Design simple components on the basis of knowledge of stress, strain and strength of material.							

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO			
1	Fundamental Concepts and Definitions	oncepts and Specific volume, Pressure, temperature. Thermodynamic equilibrium, Property,					
2	First law Second law	First law of thermodynamics, Concept of processes, Flow processes and control volume, Flow work, Steady flow energy equation, Mechanical work in a steady flow of process.  Essence of second law, Thermal reservoir, Heat engines, COP of heat pump and refrigerator. Statements of second law, Carnot cycle, Clausius inequality.	08	CO2			
3	Basic Concept Friction	Laws of motion, Transfer of force to parallel position, Resultant of planer force system. Free Body diagrams, equilibrium and its equation.		CO3			
4	Structure analysis	Structure Beams: Introduction, Shear force and bending moment, Shear and bending					
5	Stress and strain analysis	Simple Stress and strain: Introduction, Normal, shear stresses, Stress-strain diagrams for ductile and brittle materials.  Pure Bending of Beams: Introduction, Simple bending theory.	08	CO5			

#### **Reference Books:**

Van Wylen G.J. & Sonnlog R.E. Fundamentals of Classical Thermodynamics, John Wiley & Sons, Inc. NY.

Wark Wenneth: Thermodynamics (2nd edition) Mc Graw Hill Book Co. NY.

Holman, J.P.: Thermodynamics, Mc Graw Hill Book Co.NY.

Shames I.H., Engineering Mechanics, P.H.I.

D.S. Kumar, Mechanical Engineering, S.K. Katarial & Sons.

Bhavi Katti S.S., Engineering Mechanics, New Age Pub.

P.K. Bharti: Engineering Mechanics, Kataria and Sons.

#### e-Learning Source:

 $https://www.youtube.com/watch?v=Dy2UeVCSRYs\&list=PL2\_EyjPqHc10CTN7cHiM5xB2qD7BHUry7$ 

https://www.youtube.com/watch?v=DzyIEz3dKXQ&t=1s
https://www.youtube.com/watch?v=A-3W1EbQ13k&list=PLyqSpQzTE6M_MEUdn1izTMB2yZgP1NLfs

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	PO2	PO3	PO4	PU5	POO	PO/	PU	PO9	POIU	POII	PO12	P501	PS02
CO1	3	2	2	1	0	2	0	0	0	0	0	3	3	2
CO2	3	3	3	2	0	3	0	0	0	0	0	3	3	3
CO3	3	3	3	2	0	3	0	0	0	0	0	3	3	2
CO4	3	2	2	2	0	3	0	0	0	0	0	3	3	2
CO5	3	3	2	1	0	3	0	0	0	0	0	3	3	2

Name & Sign of Program Coordinator	Sign & Seal of HoD



Effective from Session: 2020-21							
Course Code	CS101	Title of the Course	Computer Programming	L	T	P	C
Year	I	Semester	I	3	1	0	4
Pre-Requisite	Nil	Co-requisite	Nil				
Course Objectives	<ul><li>To pr</li><li>To sh</li><li>To st</li></ul>	ovide fundamental concepts now the use of functions and udy the implementation of a	, networks, algorithms & flowcharts. s of programming language "C". l pointers to different problems. urrays, matrices and strings. data types structure & union.				

	Course Outcomes								
CO1	inderstand basic concepts of computer, networks and formulation of algorithmic solutions to problems.								
CO2	Understanding of programming concepts of C language and their implementation.								
CO3	Analyze and develop programs on pointers and functions.								
CO4	Develop programs on different operations on arrays, matrices & strings.								
CO5	Implement programs on structure, union & Dynamic memory allocation.								

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Introduction to Computers	Generation of computers, Characteristic and classifications of computers. Components of Computer: CPU, Various I/O Devices, Memory & its types, (MemoryHierarchy, Storage Media), Computer Software and their types, Operating System. Computer Networks & Communication: LAN, MAN, WAN, Network Topologies, Modes of Data Communication. Introduction to Internet and its Safeguard: Internet Addresses, Domain Name System, URL, Web Browsers Search Engines, Firewalls, Anti-Virus, Translators. Algorithm and flowchart: Algorithm and flow chart characteristics, Sketching Flowcharts of various problems.	08	COI
2	Starting C	Standard I/O in "C", "C" Fundamental, C Character set, Constants, Variables, Keywords and Identifiers, Data types, Declaration. Operators and Expressions, Conditional statements (If, If-else), Nesting of if-else statement, switch statement, The? operator, goto statement. Decision making and Looping (While, Do-While, for), Break and Continue statements, Case Control Structures (Switch), C programs based on above concepts.	08	CO2
3	Introduction to pointers	Declaration and initialization of pointers, accessing the address of the variable, accessing the variable through the pointer, chain of pointers, pointers operators, pointer arithmetic Introduction to Functions: Need of "C" function, User Defined and Library Functions, Prototype of Function, Call by Value; Call by Reference; Nesting of Functions, Recursion. Pointers with function, C program based on above concept.	08	CO3
4	Array	Concept of One Dimensional and Multi-Dimensional arrays, Declaration, Operations: insert, delete, search, traverse, and merge, matrix operations, Sorting: Bubble sort, merge sort, insertion sort. Character array and strings: declaring and initializing strings variable, reading and writing a character, reading and writing strings from terminal, Arithmetic operations on characters, string handling functions. Application of pointers, and function on array, C program based on above concept.	08	CO4
5	Structures	Defining Structure, Declaration of Structure Variable, Accessing Structure members, copying and comparing structure variable, operation on individual member, nesting of structures, Array of structures. Application of pointers and function on Structures. Union Defining Union Declaration of Union, difference between structure and Union, Introduction of Static and Dynamic memory allocation- The process of Dynamic memory allocation, C program based on above concept.	08	CO5
Referen	nce Books:			

Foundation of Information Technology by D.S. Yadav"- New age International						
Programming in "C" by E Balagurusamy"TMH Publication.						
Let us "C" by "Yashwant Kanitkar"-BPB Publication.						
The C Programming Essentials by Dey- Pearson Publication.						
e-Learning Source:						
https://onlinecourses.nptel.ac.in/noc22_cs40/preview						

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	101	102	103	104	103	100	107	100	109	1010	ron	FO12	1301	1302
CO1	3	2	0	0	0	3	0	3	2	0	0	2	3	2
CO2	3	3	1	0	0	1	0	2	0	0	0	2	1	3
CO3	3	2	0	0	0	2	3	2	0	0	0	3	0	0
CO4	3	2	0	2	0	3	2	2	0	0	0	1	0	0
CO5	0	0	1	0	0	1	0	0	0	0	0	0	1	0

Name & Sign of Program Coordinator	Sign & Seal of HoD



Effective from Session: 2015-16								
Course Code	ES101	Title of the Course	Title of the Course Environmental Studies					
Year	I	Semester	I	2	1	0	3	
Pre-Requisite	10+2 with Physics, Chemistry & Biology/ Math's	Co-requisite	Nil					
Course Objectives	ecosystem. This will help st	tudents in enhancing thei	rt basic and key knowledge of envir knowledge of biodiversity and it will able to explore concept of the s	s con	serva	ition.	r	

	Course Outcomes							
CO1	Gain knowledge about environment and ecosystem							
CO2	Students will learn about natural resource, its importance and environmental impacts of human activities on natural resource.							
CO3	Gain knowledge about the conservation of biodiversity and its importance.							
CO4	Aware students about problems of environmental pollution, its impact on human and ecosystem and control measures.							
CO5	Students will learn about increase in population growth and its impact on environment.							

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Introduction to Environment and Ecosystems	Environment, its components and segments, Multidisciplinary nature of Environmental studies, Concept of Sustainability and sustainable development, Environmental movements, Ecosystem, Structure & Function, Energy flow in the Ecosystem, Ecological Pyramids and Ecological Succession.	08	CO1
2	Natural Resources	Renewable and non-renewable, Soil erosion and desertification, Deforestation, Water: Use and over exploitation, Impacts of large Dams, Case studies	08	CO2
3	Biodiversity and Conservation	Levels of biological diversity, Hot spots of biodiversity, India as a Mega Diversity Nation, Endangered and endemic species of India, Threats to Biodiversity, Conservation of Biodiversity, Ecosystem and biodiversity services.	08	CO3
4	Environmental Pollution, Policies and Practices	Environmental pollution, Solid waste management, Ill effects of fireworks, Climate change, Ozone layer depletion, acid rain and impacts on human communities and Environment, Environmental Laws: Environment Protection Act, Wildlife protection Act, Forest conservation Act, Convention on Biological Diversity (CBD), Tribal rights, Human wildlife conflicts.	08	CO4
5	Human Population and the Environment	Human population growth: Impacts on environment, human health and welfare, Resettlement and rehabilitation of project affected persons, Environmental ethics, Environmental communication and public awareness, case studies.	08	CO5

#### **Reference Books:**

Agarwal, K.C. 2001 Environmental; Biology, Nidi Pub. Ltd. Bikaner.

Bharucha Erach, The Biodiversity of India, Mapin Pub. Pvt. Ltd., Ahemdabad-380, India.

Brunner R.C. 1989. Hazardous waste incineration, Mc Graw Hill

Clark R.S. Marine Pollution, Clanderon Press Oxford (TB)

Cunningham W.P.2001.Cooper, T.H. Gorhani, E & Hepworth, Environmental encyclopedia, Jaicob Publication House, Mumbai.

De. A.K. Environmental chemistry Willey Eastern Limited.

Glick, H.P.1993 water in crisis, Pacific Institute for studies in dev, Environment & security, Stockholm Env, Institute, Oxford Univ, Press 473 p.

Hawkins R.E. Encyclopedia of Indian Natural History, Bombay Natural History Society, Bombay.

Heywood, V.H. & Watson, R. T.1995.Global biodiversity Assessment .Cambridge Univ. Press 1140 p.

Jadhave, H. and Bhosale, V. M. 1995 Environmental protection and laws, Himalaya pub, house, Delhi.284 p.

e-Learning Source:
https://byjus.com/biology/difference-between-environment-and-eCOsystem.
https://www.youtube.com/watch?v=dRPl4TB8w7k
https://www.youtube.com/watch?v=3fbEVytyJCk
https://www.vedantu.com/biology/conservation-of-biodiversity
https://youmatter.world/en/definition/soil-erosion-degradation-definition/
https://byjus.com/biology/difference-between-environment-and-eCOsystem.

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	101	102	103	104	103	100	100 107	100	10)	1010	FOII	FO12	1301	1302
CO1	1	2	2	1	1	1	3	1	1	1	1	2	1	1
CO2	1	1	1	1	1	1	3	1	1	1	1	1	1	1
CO3	1	1	2	1	1	1	2	1	1	1	1	2	1	1
CO4	1	1	1	1	1	1	2	1	1	1	1	2	1	1
CO5	1	1	2	1	1	2	3	2	1	2	1	2	1	1

Name & Sign of Program Coordinator	Sign & Seal of HoD



Effective from Session: 2019-20											
Course Code	CH102	Title of the Course	Engineering Chemistry Lab	L	T	P	C				
Year	I	Semester	I	0	0	2	2				
Pre-Requisite	10 + 2 with Chemistry	10 + 2 with Chemistry <b>Co-requisite</b>									
Course Objectives	Improvement of practi	<ul> <li>To understand qualitative and quantitative problems</li> <li>Improvement of practical/technical skills.</li> <li>Ability to work effectively and safely in a laboratory environment.</li> </ul>									
	Enhancing communication skill.										

	Course Outcomes							
CO1	Analysis of iron ore.							
CO2	Study of water quality parameters.							
CO3	Study of Iodometric titration.							
CO4	Comprehension of principle, instrumentation and use of UV-VIS spectrophotometer and pH meter.							
CO5	Detection of functional groups and elements in organic compounds.							

Unit No.	Title of the Experiment	Content of Unit	Contact Hrs.	Mapped CO		
1	Iron content	To determine the Iron content in the given iron ore by using external indicator.	02	1		
2	Alkalinity	To determine the Alkalinity in the given water sample.	02	2		
3	Chloride content	To determine the Chloride content in the given water sample by Mohr's method. (Argentometric method).	02	2		
4	Available chlorine	To determine the Percentage of Available Chlorine in the given				
5	Hardness	To determine the temporary and permanent hardness in water sample by Complexometric titration using EDTA as standard solution.	02	2		
6	Chemical displacement	To determine the Equivalent weight of Iron by Chemical Displacement method. (The Equivalent weight of copper is 63.5)	02	3		
7	pH metric determination	To determine the strength of given HCl solution by titrating it against NaOH solution using pH meter.	02	4		
8	Spectrophotometric measurement	To determine the iron concentration in the given water sample by Spectrophotometer using potassium thiocyanate as color developing agent.	02	4		
9	Functional group detection			5		
10	Elements detection	·				

#### **Reference Books:**

Fundamentals of Chemistry with Quantitative analysis-I, R.L. Madan., S.Chand Publications

Advance Practical Chemistry: Jagdamba Singh, L.D.S Yadav, Jaya Singh, I.R. Siddiqui, PragatiEdition.

Practical Organic Chemistry, A.I.Vogel.

#### e-Learning Source:

https://www.bing.com/videos/search?q=functinal+group+detection&&view=detail&mid=F232CD67537BBA0CC3EBF232CD67537BBA0CC3EB&&FORM=VRDGAR&ru=%2Fvideos%2Fsearch%3Fq%3Dfunctinal%2520group%2520detection%26qs%3Dn%26form%3DQBVR%26%3D%2525eManage

https://www.bing.com/videos/search?q=alkalinility+of+water+sample&qpvt=alkalinility+of+water+sample&view=detail&mid=7AF6506DB69D2C2F3EA37AF6506DB69D2C2F3EA3&&FORM=VRDGAR&ru=%2Fvideos%2Fsearch%3Fq%

https://www.bing.com/videos/search? q=iodometric+titration&qpvt=Iodometric+titration&FORM=VDRE

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	POZ	103	PO4	PU5	roo	ro/	rus	ros	PO10	POII	POIZ	PS01	PSU2
CO1	3	2	0	0	0	0	2	1	2	2	0	2	0	1
CO2	3	2	2	0	0	0	2	1	2	2	0	2	1	-
CO3	3	2	0	0	0	0	2	1	2	2	0	2	0	1
CO4	3	2	0	2	1	0	2	1	2	2	0	2	0	1
CO5	3	2	0	0	0	0	2	1	2	2	0	2	0	1

1-Low Correlation; 2- Moderate Correlation; 3- Substantial Correlation

Name & Sign of Program Coordinator Sign & Seal of HoD



Effective from Session: 2015-16									
Course Code	ME102	Title of the Course	Mechanical Engineering Lab	L	T	P	C		
Year	I	Semester	I	0	0	2	1		
Pre-Requisite	NIL Co-requisite NIL								
Course Objectives	<ul> <li>through stu</li> <li>To underst refrigeratio</li> <li>To understa</li> <li>To learn the</li> </ul>	dy their models.  and the working and ba n system through model st and basic components and e technique for determine	sic components of 4 stroke petrol engine and 4 stroke components of 2 stroke petrol and valudy.  working of water tube boiler through model of hardness and impact strength of a material of compressive strength of a brick through U	npor study	comj				

	Course Outcomes							
CO1	To understand the working of 4 stroke petrol engine and 4 stroke Diesel engine through study their models.							
CO2	To understand the working of 2 stroke petrol and vapor compression refrigeration system through model study							
CO3	To understand basic components and working of water tube boiler through model study.							
CO4	To learn the technique for determine of hardness and impact strength of a material.							
CO5	To learn the technique for determine of compressive strength of a brick through UTM.							

Experiment No.	Title of the Experiment	Content of Unit	Contact Hrs.	Mapped CO
1	Four Stroke Petrol Engine	To Study & Sketch the model of S.I. Engine (4 Stroke)	02	CO1
2	Four Stroke Diesel Engine	To Study & Sketch the model of C.I. Engine (4 Stroke).	02	CO1
3	Two Stroke Petrol Engine	To Study & Sketch the model of S.I. Engine (2 Stroke)	02	CO2
4	Vapor Compression	To Study & Sketch the model of Vapor Compression Refrigerators	02	CO2
5	Water Tube Boiler	To Study & Sketch the model of water tube boiler (Babcock & Wilcox)	02	CO3
6	Impact Testing	To determine the Impact Strength of Mild Steel using Izod Method	02	CO4
7	Hardness Testing	To determine the harness of a mild steel specimen by using hardness tester (Rockwell Hardness test)	02	CO4
8	UTM Testing	To learn the technique for determine of compressive strength of a brick through UTM.	02	CO5
e-Learning	Source:			
https://www.	vlab.co.in/			

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	102	103	104	103	100	107	100	109	1010	1011	1012	1301	1502
CO1	3	2	2	1	0	3	0	0	3	2	0	3	3	2
CO2	3	2	2	2	0	3	0	0	3	2	0	3	3	2
CO3	3	2	3	2	0	3	0	0	3	2	0	3	3	2
CO4	3	3	3	2	0	3	0	0	3	2	0	3	3	2
CO5	3	3	2	1	0	3	0	0	2	2	0	3	3	2

1-Low Correlation; 2- Moderate Correlation; 3- Substantial Correlation

Name & Sign of Program Coordinator	Sign & Seal of HoD



Effective from Session: 2020-21									
Course Code	CS102	Title of the Course	of the Course Computer Programming Lab L T						
Year	I	Semester	I	0	0	2	1		
Pre-Requisite	None	Co-requisite	None						
Course Objectives	<ul><li>To be able to</li><li>To learn the</li><li>To learn the</li></ul>	use of C libraries functions file handling and basic men	them to create programs and applications us		langua	ge.			

	COURSE OUTCOME							
CO1	Able to understand the basic concepts of C programming language and their implementation.							
CO2	Able to design and develop various programming problems using C programming concepts.							
CO3	Able to analyze and develop programs on pointers and functions.							
CO4	Able to develop programs on different operations on arrays, matrices & strings.							
CO5	Able to implement programs on structure, union & Dynamic memory allocation.							

Experiment No.	Title of the Experiment	Content of Unit	Contact Hrs.
1	Write a Program to print any message.	1	1
2	Write a Program to print sum and multiply of two numbers.	1	1
3	Write a Program to enter the temperature in Celsius(c) then count it into Fahrenheit.	1	1
4	Write a Program to swap the number taking the help of third variable.	1	1
5	Write a Program to calculate the volume of box.	1	1
6	Write a Program to swap the number without taking the help of third variable.	2	2
7	Write a Program to check a year is leap year not.	2	2
8	Write a Program to print number is even or odd.	2	2
9	Write a Program to Print month of name using switch case.	2	2
10	Write a Program to print the no is positive or negative.	2	2
11	Write a Program to find the greater number enter by user.	2	3
12	Write a Program to find the greater number Input 3 No.	2	3
13	Write a Program to enter any no and check whether the given no is palindrome or not.	2	3
14	Write a Program to enter any no. and check whether the given no. is Armstrong or not.	2	3
15	Write a Program to Print Pattern  *  **  ***	1	3
16	Write a Program to Print Pattern1 2 3 4 1 2 3 1 2 1	1	4
17	Write a Program to Print Pattern1 1 2 1 2 3 1 2 3 4	1	4
18	Write a program to find in C to design the report card of 5 subject according to the following condition if the totalpercentage are.  >=35 and <45 IIIrdDiv  >=45 and <60 IIndDiv  >=60 IstDiv  If any students score <35 in any of the subject display fail	1	4
19	Write a Program to create 2-D array or order M*N and insert the element and display it.	2	4

20	Write a Program to find the addition of two matrix of order M*N.	2	4
21	Write a Program to find the Transpose of the matrix.	2	5
22	Write a Program to swap two numbers Call by Value.	2	5
23	Write a Program to swap two number using function pointers.		5
24	Write a Program for structure of player Name, batting average and then name.	2	5

#### **Reference Books:**

Foundation of Information Technology by 'D.S. Yadav'

Programming in 'C' by 'E Balagurusamy'.

3Let us 'C' by 'YashwantKanitkar'

The C Programming Essentials by Dey

#### e-Learning Source:

https://onlinecourses.nptel.ac.in/noc22\_cs40/preview

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	101	102	103	104	103	100	107	100	109	1010	FOII	FO12	1301	1302
CO1	1	1	2	0	3	0	3	0	0	0	0	0	2	1
CO2	1	1	1	2	1	0	3	0	0	0	0	0	2	1
CO3	1	2	2	2	0	0	3	0	0	0	0	0	2	1
CO4	1	2	2	2	0	0	3	0	0	0	0	0	2	1
CO5	1	2	1	0	0	0	3	0	0	0	0	0	2	1

Name & Sign of Program Coordinator	Sign & Seal of HoD



<b>Effective from Session:</b>	2020-21						
Course Code	LN151 Title of the Course Basic Professional Communication Lab					P	C
Year	I	Semester	I	0	0	2	1
Pre-Requisite	10+2	Co-requisite	U.G. Program				
Course Objectives	Ianguage The key commun The Dep expertise for Spec Students commun which do	e for professional purpose component of the varication in the English la partment of Languages e and excellence in profesific/Special Purposes (E will be given new dication to boost their coepends not only on the horse will help them overce	e students in both the artistry and utility of es through the study of language and literature ious types of professional communication inguage which is now a global language. caters to the needs of the students aspiring essional communication with a marked emphases. SP). insights into the concepts of soft skills infidence which will help them choose and be ard skills, but on one's soft skills & profession ome their fear & anxiety of public speaking whom people love to hear.	re. is ba g for asis or s & uild a onal et	training training English Engl	ng, ish sional career	r

	COURSE OUTCOME								
CO1	Students will be introduced to the basic understanding of communication and Professional Communication. Knowledge of Professional, cultural and cross-cultural communication will be imparted. Meaning and process of communication, verbal and nonverbal communication will be focused.  Basic Understanding of communication and Professional/Business Communication will be provided. They will alsolearn & practice how to introduce oneself in professional setting & how to manage speaking anxiety.								
CO2	Corrections in basic English sounds and correct pronunciations will be practiced by various Listening exercises & word games to help them become better conversationalist.								
СОЗ	Basic tools of communication and improvement in communicative competence.								
CO4	Phonetic Alphabet and Phonetic Transcriptions will be taught & practiced to improve vocal clarity & pronunciation.  Understanding the structural and functional grammar and basic structure of language.								
CO5	Intonation & Stress will be practiced to make them learn how paralinguistic features dramatically affect meaning & how it can help one in becoming a persuasive & engaging speaker.								

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Introduction	Difference between Introduction and Description, SWOT Analysis	06	CO 1
2	Software -I	Listening exercises, Pronunciation improvement through self- testing, Vocabulary improvement through word games	06	CO 2
3	Software – II	Conversational skills, Exercises based on Language Skills/ Small talk, Cultural movies	06	CO 3
4	Phonetics	Phonetic Alphabet and Phonetic Transcriptions	06	CO 4
5	Non-verbal communication	Intonation and Stress	06	CO 5

#### Reference Books:

Gerson, Sharon J. Technical Writing: Process and Product ( $5^{th}$  edition). Prentice Hall, 2005.

K. Floyd, Interpersonal Communication: The Whole Story. McGraw Hill, 2009.

Greenbaum, Sidney and Nelson Gerald, An Introduction to English Grammar. Routledge, 2009.

Swan, Michael, Practical English Usage. OUP, 2005.

Murphy, Raymond. English Grammar in Use. Cambridge University Press, 2019.

Kumar, Sanjay and Pushp Lata., Communication Skills. Oxford University Press, Oxford 2011.

Gerson, Sharon J. Technical Communication: Process and Product (9th edition). Longman Pub., 2016.

#### e-Learning Source:

https://ndl.iitkgp.ac.in./

https://epgp.inflibnet.ac.in/Home/ViewSubject?catid=9RA537jM1m7VD3VCoav4lQ==

https://library.iul.ac.in/

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	101	102	02   103	104	105	PO6	107	100	10)	1010	1011	FO12	1301	1302
CO1	3	3	3	3	3	3	3	0	0	0	0	0	3	3
CO2	3	3	3	3	3	3	3	0	0	0	0	0	3	3
CO3	3	3	2	3	3	3	3	0	0	0	0	0	3	3
CO4	3	3	2	3	3	3	3	0	0	0	0	0	3	3
CO5	3	3	3	3	3	3	3	0	0	0	0	0	3	3

Teaching and Learning methods	Flipped Classrooms, Concept Mapping, Information-Based Approach, Personal-Response Approach, Language-Based Approach, Paraphrastic Approach, Moral- Philosophical Approach and Stylistics Approach
List/Topics/Activities Planned that are beyond Syllabus	Information-Based Activities, Personal-Response Activities, Language-Based Activities, Periphrastic Activities, Moral-Philosophical Activities, and StylisticsActivities, Presentations, Small talk, Situational talk, role playing, Group Discussions, Assignments are used as a medium to work on cognitive development/growth.

Name & Sign of Program Coordinator	Sign & Seal of HoD



Effective from Session: 2015-16									
Course Code	PY101	Title of the Course	Physics	L	T	P	C		
Year	I	Semester	II	3	1	0	4		
Pre-Requisite	10+2 with Physics and Mathematics	Co-requisite							
Course Objectives		graduate course is to impar y for a strong engineering l	t basic knowledge of fundament knowledge base.	al cor	ncept	of			

	Course Outcomes
CO1	To analyze the connection between daily life observations and science.  To realize that apparently different ideas of Optics such as Interference and Diffraction have interrelationship between them.  To realize the simplicity of ideas involved in explaining complex phenomenon.
CO2	To grow in ideas of different aspect of light and develop connection between daily life applications and science. To analyze the process of development of a new theory while dealing with Polarization.  To correlate that the conceptualization of an idea is far ahead than its practical realization while dealing with LASER. To grow in realization of totally different manifestation of light.  To find the most recent applications of light in terms of communication and storage of data. To realize that how the design of complex systems is based on the simple ideas.  To realize that the conceptualization of an idea is far ahead than its practical realization while dealing with Optical Fibers.
CO3	To grow in developing connection between philosophy and science.  To find that seemingly different ideas such as Optics and Mechanics have interrelationship between them. To understand the process of development of a new theory and its application in life.  To realize the requirement of power of imagination.
CO4	To grow in developing the connection between philosophy and science.  To find that seemingly different ideas such as Compton Effect and Quantum Theory have interrelationship between them.  To understand and analyze the process of development of a new theory and how the development of one idea leads to the development of a apparently different idea.  To realize and appreciate the efforts made by the individuals to give a new understanding of science that led to the modern day applications.
CO5	To grow in developing connection between daily life utility and material science.  To realize that apparently different materials with respect to Electric and Magnetic properties have inter relationship between them. To evaluate that how totally different manifestation of Modern Science leads to new technology.  To do the evaluation that how an idea is far ahead than its practical realization while dealing with Nano Technology and Super Conductivity.

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Wave Optics	Methods of formation of coherent sources, Fresnel's Bi-prism, displacement of fringes, thin film interference, Newton's ring. Fraunhoffer diffraction at single slit, grating, Rayleigh's criterion of resolution, resolving power of grating.	08	CO1
2	Optical activity andModern Optics	Production of plane polarized light by reflection and Double refraction, Nicol prism, Optical activity, polarimeter(Laurent's and Bi-quartz). Principle of fiber optics, numerical aperture, attenuation, dispersion in optical fibers, material dispersion, waveguide dispersion, intermodal and intra-modal dispersion, Pulse dispersion in step index fiber. Main components of laser, Einstein's coefficients, He-Ne laser, Nd-YAG laser and their applications.	08	CO2
3	Properties of Matterand Relativistic Mechanics	Viscosity, Poiseulli's equation, Michelson-Morley experiment and its implications, Galilean transformation equations, Lorentz transformation equations and their consequences, energy mass relation, relativistic kinetic energy.	08	CO3
4	Quantum Physics	Compton effect, basic postulates of quantum mechanics, Wave function and its physical admissibility, orthogonality of wave functions, normalization of wave functions, Heisenberg's uncertainty principle (no derivation) and its applications (non-existence of electron in nucleus, Bohr's radius), Schrodinger's equation and its application to free particle, particle in one dimensional box	08	CO4
5	Physics of Materials	Magnetic Properties: Magnetization, Origin of magnetic moment, dia, para and ferro magnetism, Langevin's theory for diamagnetic material, Phenomena of hysteresis and its applications. Superconductors: Temperature dependence of resistivity in superconducting materials, Effect	08	CO5

of magnetic field (Meissner effect), Temperature dependence of critical	
field, Type I and Type II superconductors, BCS theory (Qualitative), High	
temperature superconductors and Applications of Super-conductors. Nano-	
Materials: Basic principle of nanoscience and technology, structure,	
properties and uses of Fullerene and Carbon nanotubes, Applications of	
nanotechnology.	

#### **Reference Books:**

Fundamentals of Optics by Jenkins and White

Optical Fiber Communication by Gerd Keiser

Concepts of Modern Physics by Arthur Beiser

Introduction to Special Theory of Relativity by Robert Resnick

Quantum Physics by Eisberg

Introduction to Nanotechnology by Poole Owens, Wiley India

Solid State Physics by S.O. Pillai, New Age Publications

#### e-Learning Source:

https://nptel.ac.in/courses/115/101/115101011/

https://nptel.ac.in/courses/115/107/115107095/

https://nptel.ac.in/courses/113/106/113106093/

https://nptel.ac.in/courses/115/101/115101107/

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	101	102	103	PO4	105	PO6	ro/	100	10)	1010	ron	FO12	1301	PS02
CO1	0	3	2	1	3	3	1	3	2	2	1	3	2	1
CO2	3	3	2	1	3	3	1	2	2	3	2	3	2	1
CO3	3	3	3	3	1	3	1	3	2	2	1	3	2	1
CO4	3	3	3	2	2	3	1	2	2	1	1	3	2	1
CO5	3	3	3	3	3	3	3	2	2	2	3	3	2	1

Name & Sign of Program Coordinator	Sign & Seal of HoD



<b>Effective from Session:</b> 2	2020-21						
Course Code	LN101	Title of the Course	Basic Professional Communication	L	T	P	C
Year	I Semester II		2	1	0	3	
Pre-Requisite	Nil	Co-requisite	Nil				
Course Objectives	<ul> <li>profess</li> <li>The ke in the l</li> <li>The Do excelled</li> </ul>	sional purposes through to by component of the vari- English language which is epartment of Languages	e students in both the artistry and utility of the he study of language and literature. ous types of professional communication is bas s now a global language. caters to the needs of the students aspiring for t munication with a marked emphasis on English	ically rainir	comm	unicati	ion ınd

	COURSE OUTCOME
CO1	Students will be introduced to the basic understanding of communication and Professional Communication.  Knowledge of Professional, cultural and cross-cultural communication will be imparted. Meaning and process of communication, verbal and nonverbal communication will be focused.
CO2	Learning Language through literature aims to develop the students' ability to read the prescribed essays and stories critically and to understand the historical-political and cultural dynamics underlying them.
CO3	Basic tools of communication and improvement in communicative competence.
CO4	Understanding the structural and functional grammar and basic structure of language.
CO5	Enhancement of writing skills in English i.e., writing application, report and various types of letters.

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Professional Communication	Professional Communication: Its Meaning and Importance, Essentials of Effective Communication, Barriers to Effective Communication	08	CO 1
2	Language through Literature	<ul> <li>A. Essays:</li> <li>1. The Effect of Scientific Temper on Man by Bertrand Russell</li> <li>2. The Aim of Science and Humanities by Moody E. Prior</li> <li>B. Short Stories:</li> <li>1. The Meeting Pool by Ruskin Bond</li> <li>2. The Portrait of a Lady by Khushwant Singh</li> </ul>	08	CO 2
3	Basic Vocabulary	Euphemism, One-word Substitution, Synonyms, Antonyms, Homophones, Idioms and Phrases, Common Mistakes, Confusable Words and Expressions, Portmanteau Words, Foreign Words and Expressions.	08	CO 3
4	Basic Grammar	Articles, Prepositions, Tenses, Concord, (Subject-Verb agreement), Modal Auxiliaries, Verbs: its Kinds and uses, Degrees of Comparison, Punctuation	08	CO 4
5	Basic Composition	Report Writing: What is report? Kinds and Objectives of reports, writing reports, Business Letter writing; Introduction to Business Letters, Layout of Business letters, Letters of Enquiry/Complaint Proposal writing	08	CO 5

#### **Reference Books:**

Gerson, Sharon J. Technical Writing: Process and Product (5th edition). Prentice Hall, 2005.

K. Floyd, Interpersonal Communication: The Whole Story. McGraw Hill, 2009.

Greenbaum, Sidney and Nelson Gerald, An Introduction to English Grammar. Routledge, 2009.

Swan, Michael, Practical English Usage. OUP, 2005.

Murphy, Raymond. English Grammar in Use. Cambridge University Press, 2019.

Kumar, Sanjay and Pushp Lata., Communication Skills. Oxford University Press, Oxford 2011.

Raman, Meenakshi, and Sangeeta Sharma. Technical Communication: Principals and Practice. Second Edition, Oxford University Press, 2012.

Gerson, Sharon J. Technical Communication: Process and Product (9th edition). Longman Pub., 2016.

#### e-Learning Source:

http://www.uptunotes.com/notes-professional-communication-unit-i-nas-

https://www.docsity.com/en/subjects/professional-communication/
https://lecturenotes.in/download/note/22690-note-for-communication-skills-for-profession

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	PO2	PO3	PO4	105	POO	PO/	PU	PO9	POIU	POII	PO12	P501	P502
CO1	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO2	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO3	3	3	2	3	3	3	3	3	3	3	3	3	3	3
CO4	3	3	2	3	3	3	3	3	3	3	3	3	3	3
CO5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

Teaching and learning methods	Flipped Classrooms, Concept Mapping, Information-Based Approach, Personal-Response Approach, Language-Based Approach, Paraphrastic Approach, Moral-Philosophical Approach and Stylistics Approach
List/Topics/Activities Planned that are beyond Syllabus	Information-Based Activities, Personal-Response Activities, Language-Based Activities, Periphrastic Activities, Moral-Philosophical Activities, and Stylistics Activities

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Name & Sign of Program Coordinator	Sign & Seal of HoD



<b>Effective from Session:</b>	Effective from Session: 2017-18							
Course Code	MT112	Title of theCourse	Engineering Mathematics - II	L	Т	P	C	
Year	I	Semester	II	3	1	0	4	
Pre-Requisite	10+2 Mathematics	Co- requisite	Nil					
Course Objectives		The course is aimed to develop the skills in mathematics which is necessary for grooming them into successfulengineering graduate. The topics introduced will serve as basic tools for specialized studies in science field.						

	COURSE OUTCOME
CO1	Solve first order linear equations and higher order differential equation of certain types and interpret the solutions.
CO2	To use shift theorems to compute the Laplace transform, inverse Laplace transform and the solutions of second order, linear equations with constant coefficients.
CO3	Able to determine given function in terms of sine and cosine terms in Fourier series.
CO4	Apply problem-solving using concepts and techniques from PDE's and Fourier analysis applied to diverse situations in physics, engineering, financial mathematics and in other mathematical contexts.
CO5	Apply method of least squares to find the curve of best fit for the given data

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Unit I	Linear differential equations of first order, Linear differential equations of higher order with constant coefficients, Complementary functions and particular integrals, Solution of second order differential equations by changing dependent and independent variables, Method of variation of parameters, Applications to engineering problems (without derivation).	08	CO 1
2	Unit II	Laplace transform of different types of functions, Laplace transform of derivatives and integrals, Unit step function, Laplace transform of periodic functions, Inverse Laplacetransform, Convolution theorem, Applications to solve simple linear differential equations.	08	CO 2
3	Unit III	Periodic functions, Trigonometric series , Fourier series of period 2 $^{\pi}$ , Euler's formulae, Even and odd functions, Functions having arbitrary period, Change of interval, Half range sine andcosine series Introduction of partial differential equations, Solution of second order linear homogeneous partial differential equations with constant coefficients and their classifications to parabolic, elliptic and hyperbolic forms with illustrative examples.	08	CO 3
4	Unit IV	Method of separation of variables for solving partial differential equations, Wave equation up to two-dimensions, Heat conduction equations up to two dimensions, Laplace equation.	08	CO 4
5	Unit V	Mean, Median, Mode, Standard deviation and Variance, Method of least squares, Curve fittingof straight line and parabola.	08	CO 5
Referen	ice Books:			

#### **Reference Books:**

Advanced Engineering Mathematics, Wiley Eastern Ltd.

Advanced Engineering Mathematics, Khanna Publication.

3Higher Engineering Mathematics, Khanna Publication.

Advanced Engineering Mathematics, CBS Publication.

#### e-Learning Source:

https://nptel.ac.in/courses/111106100/

https://nptel.ac.in/courses/111105123/

 $https://courses.maths.ox.ac.uk/node/view\_material/1720$ 

https://nptel.ac.in/courses/111103021/

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	PO2	PO3	PO4	PU5	PO0	PO/	PU	PO9	POIU	POII	PO12	P501	PSO2
CO1	3	2	1	2	2	1	0	0	0	1	0	2	1	1
CO2	3	2	1	2	2	1	0	0	0	0	0	2	1	1
CO3	3	2	1	1	1	1	0	0	0	0	0	2	1	1
CO4	3	2	1	2	3	1	0	0	0	1	0	2	1	1
CO5	3	1	1	1	2	1	0	0	0	0	0	2	1	1

1-	Low Correlation; 2- Moderate Correlation; 3- Substantial Correlation
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Effective from Session: 2020-21								
Course Code	EE103	Title of the Course	Basic Electrical Engineering	L	T	P	C	
Year	I	Semester	П	3	1	0	4	
Pre-Requisite	None	Co-requisite	None					
G 01: 4:	<ul> <li>Knowledge and concept of D.C Circuit Analysis and Network Theorems Circuit.</li> <li>Use of Steady State Analysis of Single-Phase AC Circuits AC fundamentals.</li> <li>Knowledge and concept of Three Phase AC Circuits Three phase system and measuring devices.</li> <li>Basic concepts of Power System and Transformer</li> <li>Study of Electromechanical energy conversion devices: AC/DC Machines.</li> </ul>							

	Course Outcomes
CO1	Know about the concept of D.C Circuit Analysis and Network Theorems Circuit.
CO2	Steady State Analysis of Single Phase AC Circuits AC fundamentals.
CO3	Know about concept of Three Phase AC Circuits Three phase system and measuring devices
CO4	Layout of Power System and transformer
CO5	Know about Electromechanical energy conversion devices: AC/ DC Machines

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	D.C Circuit Analysis and Network Theorems	Circuit concepts: Concept of network, Active and passive elements, linear network and non linear network, unilateral and bilateral elements, lumped and distributed network, sources, open circuit and short circuit, source transformation, Kirchhoff's Law. Loop analysis and nodal analysis, star delta transformation. Network theorems: Needs of theorem, Superposition theorem, Thevenin's theorem, Norton's theorem, maximum power transfer theorem.	08	CO1
2	Steady State Analysis of Single Phase AC Circuits	Apparent, active and reactive powers, power factor, causes and problems of low power factor, power factor improvement, resonance, bandwidth and quality factor in series circuit.	08	CO2
3	Three Phase AC Circuits	Three phase system: Its necessity and advantages, meaning of phase sequence, star and deltaconnections, balanced supply, line and phase voltage/current relationship.  Measuring Instruments: Types of instruments: construction and working principle of PMMC,MI type instruments, induction type energy meter.	08	CO3
4	Introduction ofPower System	General layout of electrical power system, standard generation, transmission and distribution voltage levels, concept of grid. Magnetic	08	CO4
5	Electromechanical energy conversion devices	DC Machines: Types, emf equation of generator and torque equation of motor, applications. Three Phase Induction Motor: Types, principle of operation, applications. Single Phase Induction Motor: Principle of operation and introduction to methods of starting, applications. Three Phase Synchronous Machines: Principle of operation of alternator, synchronous motor, applications.	08	CO5
Referen	ice Books:			

V.Deltoro, "Principle of Electrical Engg." PHI, 2009..

M.A Mallick, Dr. I. Ashraf, "Fundamental of Electrical Engg," CBS Publishers, 2010.

A. Hussain, "Basic Electrical Engg" Dhanpat Rai & sons, 2007

I J Nagrath, "Basic Electrical Engg", TMH, 2010.

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	POZ	PO3	PU4	PU5	PO0	PO/	PU	PU9	POIU	POII	POIZ	P501	PSO2
CO1	3	3	2	1	1	3	0	0	0	0	0	3	3	3
CO2	3	3	3	2	1	1	0	0	0	0	0	2	3	2
CO3	3	2	1	1	2	2	3	0	0	0	0	3	2	2
CO4	3	2	2	2	3	3	0	0	0	0	0	2	3	2
CO5	3	1	1	1	1	2	1	0	0	0	0	2	3	2
	1-Low Correlation; 2- Moderate Correlation; 3- Substantial Correlation													

1-Low Correlation: 2- Moderate Correlation: 3- Substantial Correla
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Effective from Sessi	on: 2017-18						
Course Code	EC101	Title of the Course	Basic Electronics	L	T	P	C
Year	I	Semester	П	3	1	0	4
Pre-Requisite	Nil	Co-requisite	Nil				
Course Objectives	Electrons Level. To Breakdov To under configura To under Working characteri To under Complim function u To under	and holes in intrinsic Too learn the working and on mechanism. Tolearn the restand NPN Transistor, tion andtheir characteristic stand JFET: Construction of Enhancement and depistics.  stand Switching theory & ents, Addition and Subtrasing Karnaugh map stand Operational Amplific	nanism of conduction in semiconductors: Mobile learn the semiconductors, Donor and accepit it's characteristic of PN junction diode, Zeworking half wave rectifier, full wave rectifier. Common Emitter, Common Base and Cocs, transistor biasing circuits.  In principle of working and its characteristics. In the learn Number system, Converced gates. To learn Number system, Converced gates. To learn Logic gates, Miniter. To learn Ideal characteristics of Op-Amp & g, integrator and differentiator. Block diagratator and CRO.	tor in lener ers an mmod To 1 MOS ersion mizat	npuriti and A d LED n Col earn N SFET  1, 2's a ion of	es, FeAvalar  O.  lector  MOSF  and the logic  ion, O	ET: heir )'s cal

	COURSE OUTCOME
CO1	Understand the concepts of concepts of mechanism of conduction in semiconductors. Students shall be able to draw I-V characteristic of different diodes also know the working and the applications of different diodes
CO2	Understand the concepts of NPN Transistor, Common Emitter, Common Base and Common Collector configuration and their characteristics. Students can do circuit analysis with different methods of transistor biasing.
CO3	Understand the concepts of JFET: Construction, principle of working and its characteristics. Students shall be able to differentiate between JFEF & MOSFET, working of Enhancement and depletion type N-channel MOSFET, P-channel MOSFET and their characteristics.
CO4	For a given problem, <b>apply</b> the concept of Switching theory & Logic gates students shall be able to solve mathematical problems using 2'sand 10's compliments, minimize Boolean function using different laws and K map. Students shall be able to design combinational circuits for the given Boolean function.
CO5	Understand the concept of Operational Amplifier and develop Op-Amp as an Inverting, Non-inverting, integrator and differentiator. Students are able to use digital multimeter and draw different lissajous pattern on CRO using function generator.

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Mechanism of Conduction in Semiconductors	Mobility and Conductivity, Electrons and holes in intrinsic semiconductors, Donor and acceptor impurities, Fermi Level.PN junction diode working and its characteristic, Zener and Avalanche Breakdown mechanism. Half wave and full wave rectifiers, LED.	08	CO1
2	BJT characteristics and circuits	Working of NPN Transistor, Common Emitter, Common Base and Common Collectorconfiguration and their characteristics, transistor biasing circuits	08	CO2
3	Field EffectTransistors	FieJFET: Construction, principle of working and its characteristics.  MOSFET: Working of Enhancement and depletion type N-channel  MOSFET, P-channelMOSFET and their characteristics.	08	CO3
4	Switching theory &Logic gates	Number system, Conversion, 2's and 10's Compliments, Addition and Subtraction, Boolean algebra, Logic gates, Minimization of logical function using Karnaugh map.	08	CO4
5	OperationalAmplifier	Ideal characteristics of Op-Amp ⁢'s application, Op-Amp as an Inverting, Non-inverting, integrator and differentiator. Block diagram and working of Digital Multimeter, Function Generator and CRO.	08	CO5

#### Reference Books:

Bolyested & Nashekey: Electronic Devices and Circuit Theory, PHI.

Milliman & Halkias: Integrated Electronics, McGraw-Hill.

#### e-Learning Source:

 $https://www.youtube.com/watch?v=4\_nGFY7zgDM$ 

 $https://www.youtube.com/results?search\_query=diode+characteristics\\$ 

				Course	Articul	ation M	atrix: (1	Mapping	g of CO	s with PC	s and PS	(Os)		
PO-PSO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	POZ	103	PU4	105	100	PO/	PU	PO9	POIU	POII	POIZ	PS01	PSO2
CO1	3	1	3	0	0	0	0	0	3	0	0	1	3	2
CO2	3	2	3	0	0	0	0	0	3	0	0	1	3	0
CO3	3	3	3	1	1	0	0	0	3	0	0	0	3	2
CO4	3	3	2	0	0	0	0	0	3	0	0	1	3	0
CO5	3	2	2	1	1	0	0	0	3	0	0	1	3	2

1- Low Correlation: 2- Moderate Correlation:	3- Substantial	Correlation
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Name & Sign of Program Coordinator	Sign & Seal of HoD



Effective from Session: 2015-16								
Course Code	PY104	Title of the Course	Physics Lab	L	T	P	C	
Year	I	Semester	П	0	0	2	1	
Pre-Requisite	10+2 with Physics and Mathematics	Co-requisite	Nil					
Course Objectives	1 1	his undergraduate course nents related to its theore	is to impart practical knowledge of the tical course.	conc	epts t	hroug	;h	

	Course Outcomes
CO1	To demonstrate how interference takes place by division of amplitude and by division of wavefront.
CO2	To demonstrate the practical applications of polarization phenomenon in finding the specific rotation, refractive index and Brewster's angle.
CO3	To demonstrate the practical application of Fraunh offer diffraction in wavelength and focal length calculation.
CO4	To demonstrate the magnetic and heating effect of current in finding the magnetic field and Stefan's constant.
CO5	To demonstrate how to calculate the energy band gap of a semiconductor material and viscosity of a liquid.

List of Experiments	Content of Unit
Exp.1	To determine the wave length of monochromatic light by Newton's ring.
Exp.2	To determine the wave length of monochromatic light with the help of Fresnel's Biprism.
Exp.3	To determine the focal length of two lenses by nodal slide and locate the position of cardinal points.
Exp.4	To determine the specific rotation of cane sugar solution using Half Shade polarimeter.
Exp.5	To determine the wavelength of spectral lines using plane transmission grating.
Exp.6	To determine the Brewster's angle and refractive index of material with the help of a laser source.
Exp.7	To determine the variation of magnetic field along the axis of a current carrying coil and then to estimate the radius of the coil.
Exp.8	To verify Stefan's law by electrical method.
Exp.9	To determine the energy band gap of a given semiconductor material.
Exp.10	To determine the coefficient of viscosity of a liquid.

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)													
PO-PSO	DO1	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	101	102	103	104	103	100	107	100	109	1010	ron	FO12	1301	1302	
CO1	3	3	2	1	3	1	3	0	0	0	0	0	2	1	
CO2	2	2	2	2	2	3	2	0	0	0	0	0	2	1	
CO3	3	3	1	3	3	1	3	0	0	0	0	0	2	1	
CO4	2	2	2	3	1	2	2	0	0	0	0	0	2	1	
CO5	2	1	1	1	2	2	2	0	0	0	0	0	2	1	

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Effective from Session: 2017-18												
Course Code	EE104 Title of the Course		Electrical Engineering Lab	L	T	P	C					
Year	I	Semester	П	0	0	2	1					
Pre-Requisite	NIL	Co-requisite										
Course Objectives	<ul><li>To underst</li><li>To underst</li></ul>	and and experiment with and and experiment with	n the verification of DC Network Theorems in the study of diode, rectifier, BJT character in the study of resonance and determination in the calibration of energy meter and operation	istics of tra	nsforn	ner los	ses					

	Course Outcomes
CO1	Adopt, perform, analyze and implement the methods of verification of DC Network Theorems; contribute in related development
CO2	Adopt, perform, analyze and implement the methods of study of diode, rectifier, BJT characteristics and Amplifier; contribute in related development
СОЗ	Adopt, perform, analyze and implement the methods of study of resonance and determination of transformer losses; contribute in related development
CO4	Adopt, perform, analyze and implement the methods of calibration of energy meter and operation of induction motor; contribute in related development

Unit No.	Title of the Unit	Content of Experiment	Contact Hrs.	Mapped CO
1.	Experiment -1	Verification of Thevenin's Theorem.	02	CO1
2.	Experiment -2	Verification of Superposition Theorem.	02	CO1
3.	Experiment -3	Verification of Maximum Power Transfer Theorem.	02	CO1
4.	Experiment -4	To study V-I characteristics of diode.	02	CO2
5.	Experiment -5	To study the input & output characteristics of BJT in CE configuration.	02	CO2
6.	Experiment -6	To study the full wave rectifier circuit with & without filter and determine the ripple factor.	02	CO2
7.	Experiment -7	To study the phenomenon of resonance in series RLC circuit.	02	CO3
8.	Experiment -8	Determination of losses in single phase transformer by OCT and SCT.	02	CO3
9.	Experiment -9	To calibrate a single-phase induction type energy meter.	02	CO4
10.	Experiment -10	To study the running and reversing of a three phase SCIM.	02	CO4
11.	Experiment -11	Study of OP Amp based inverting and non-inverting amplifier	02	CO2

#### **Reference Books:**

V.Deltoro, "Principle of Electrical Engg." PHI, 2009.

M.A Mallick, Dr. I. Ashraf, "Fundamental of Electrical Engg," CBS Publishers, 2010.

A. Hussain, "Basic Electrical Engg" Dhanpat Rai & sons, 2007.

R. Boylestad, "Electronic Devices and Circuit Theory", Pearson, 2013.

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)														
PO-PSO	PO1	DO1	DO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	101	102	103	104	103	100	107	100	109	1010	ron	FO12	1301	1302		
CO1	3	3	2	1	1	3	0	0	0	0	0	3	3	3		
CO2	3	3	3	2	1	1	0	0	0	0	0	2	3	2		
CO3	3	2	1	1	2	2	3	0	0	0	0	3	3	3		
CO4	3	2	2	2	3	3	0	0	0	0	0	2	3	2		

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Effective from Session: 2015-16											
Course Code	ME103	Title of the Course	Engineering Graphics	L	T	P	C				
Year	I	Semester	П	0	0	2	1				
Pre-Requisite	None	Co-requisite	None								
Course Objectives	• This • To us			e draw	ings.						

	Course Outcomes
CO1	Describe the fundamentals of engineering drawing, use of geometrical instruments and drawing steps
CO2	To understand the concept of projection and acquire visualization skills, draw the projection of points, lines and planes.
CO3	Classify solids and projection of solids at different positions
CO4	To get the exact sectioned view of solids and development of their surfaces.
CO5	To draw isometric projection and perspective views of an object.

Unit No.	Title of the Unit	Content of Unit	Contact Hrs.	Mapped CO
1	Lettering and geometrical constructions	Describe the fundamentals of engineering drawing, use of geometrical instruments and layout for initial drawing.	02	CO1
2	Orthographic projections of points	Describe the fundamentals orthographic projections and use of geometrical instruments and layout for initial drawing.	02	CO2
3	Projections of lines	Describe the fundamentals of projections of lines and use of geometrical instruments and procedure for the drawing.	02	CO2
4	Projections of solids	Describe the fundamentals of projections of solids and use of geometrical instruments and procedure for the drawing.	02	CO3
5	Sectioning of solids	Describe the fundamentals of sectioning of solids and use of geometrical instruments and procedure for the drawing.	02	CO4, CO3
6	Isometric Projections	Describe the fundamentals of Isometric projections and use of geometrical instruments and procedure for the drawing.	02	CO5
7	Production drawing	Describe the fundamentals of production drawing.	02	CO1, CO2

#### **Reference Books:**

Engineering graphics by Pradeep Jain

Engineering graphics by Krunal Patel

#### e-Learning Source:

https://www.youtube.com/watch?v=p62LPzFqGQw&list=PLp6ek2hDcoNCjoRLQ4rjpCozisCACBxKA

 $https://www.youtube.com/watch?v=VrU73IwRyc4\&list=PLLy\_2iUCG87Bw9XPfEF3r3EW5UlAOv8iz$ 

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)																			
PO-PSO	PO1	DO1	DO1	DO1	DO1	DO1	DO1	DO1	DO3	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO	POI	PO2	PO3	PO4	PU5	PO0	PO/	PU	PO9	POIU	POII	PO12	P501	PSO2							
CO1	3	0	0	0	0	0	0	0	1	2	0	3	3	0							
CO2	3	2	2	0	0	0	0	0	1	2	0	3	3	0							
CO3	3	2	2	0	0	0	0	0	1	2	0	3	3	0							
CO4	3	2	2	0	0	0	0	0	1	2	0	3	3	0							

1-Low Correlation; 2- Moderate Correlation; 3- Substantial Correlation

Name & Sign of Program Coordinator	Sign & Seal of HoD

Effective from Session: 2015-16								
Course Code	ME104	Title of the Course	Workshop Practice		T	P	C	
Year	I	Semester	II	0	0	2	1	
Pre-Requisite	None	Co-requisite	None					
Course Objectives	To impart practical knowledge and hands-on practice on the lathe machine. To impart practical knowledge of basic tools and operations in the fitting shop and carpentry shop. To impart basic knowledge of smithy tools and hands-on practice in smithy shop. To impart basic knowledge of different welding tools and equipment and hands-on practice of making different welding joints. To impart practical knowledge of different types of sheet metal tools and equipment and hands-on practice of making sheet metal components.							

Course Outcomes					
CO1	Perform different operations on lathe machine.				
CO2	Manufacture components using tools and equipments of fitting shop and carpentry shop.				
CO3	Make components in smithy shop using different types of smithy tools and equipments.				
CO4	Perform different joining operations using welding tools and equipments.				
CO5	Make sheet metal components using different sheet metal tools and equipments.				

Experiment No.	Title of the experiment	Content of Unit		Mapped CO
		To study and sketch a lathe machine		
1 Lathe machine		Practice of operations-facing, plain turning, step turning, Taper turning & chamfering	02	CO1
2	Fitting shop & carpentry shop	To study and sketch fitting tools and equipment Practice of step cutting, filing, drilling & tapping To make a 90° v-groove fitting on mild steel flat To study and sketch different types of carpentry tools & machines To make a mortise and tenon joint To make a corner lap joint	02	CO2
3	Smithy shop	To study and sketch different smithy tools & equipments To make a squire punch from mild steel round rod To make a pipe hook from a mild steel round rod	02	CO3
4	Welding shop	To study and sketch the welding equipments and tools To weld the two given plates & make a lap joint (by arc welding) To weld the two given plates & make a butt joint (by arc welding)	02	CO4
5	Sheet metal	To study and sketch different sheet metal tools & equipments To make a rectangular tray To make a conical funnel	02	CO5
e-Learning So	ource:			

## https://www.vlab.co.in/ Course Articulation Matrix: (Mapping of COs with POs and PSOs)

		Course Articulation Matrix: (Mapping of COs with POs and PSOs)												
PO-PSO	PO1	DO2	DO3	DO4	DO5	DO4	DO7	PO8	DO0	DO10	PO11	DO12	DCO1	DCO2
СО	POI	PO2	PO3	PO4	PO5	PO6	PO7	PU8	PO9	PO10	POII	PO12	PSO1	PSO2
CO1	3	2	2	3	3	2	0	0	2	0	2	3	3	2
CO2	3	2	2	2	2	2	0	0	2	0	2	3	3	2
CO3	2	2	2	2	2	2	0	0	2	0	2	3	3	2
CO4	2	2	2	2	3	2	0	0	2	0	2	3	3	2
CO5	2	2	2	2	2	2	0	0	2	0	2	3	3	2

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